Intermediate for agricultural (pesticides and fungicides) and flame retardants applications.

**SPECIFICATIONS**

- Assay, % min .......................................................... 98.5
- Iron, ppm, max .......................................................... 3
- Color, Gardner, max .................................................. 6
- Chlorides, ppm, max .................................................. 10

**PHYSICAL and CHEMICAL PROPERTIES**

- Appearance .......................................................... Dense, oily liquid
- Boiling point @ 753mm Hg ........................................... 239 °C (462.2 °F)
- Color ............................................................. Pale yellow
- Freeze Point .......................................................... 10 °C (50.0 °F)
- Molecular formula .................................................. C₅Cl₆
- Molecular weight .................................................. 273 (272.77)
- Odor ............................................................... Sharp musty odor
- Specific gravity @ 25°C ................................................. 1.702
- Vapor density @ 25°C ............................................... 9.42
- Vapor pressure, mm HG @ 20°C .................................. NDA

**SOLUBILITIES**

At 25°C, hexachlorocyclopentadiene (HEX) is soluble in acetone, carbon tetrachloride, ethanol, hexane and other organic solvents. HEX has low solubility (1.8mg/L) in water.

**HAZARDS**

**FIRE:** HEX is a non-flammable liquid. Contact with sodium may cause explosion. In case of fire, use dry chemicals, carbon dioxide, water spray or foam. Firefighters should wear self-contained breathing apparatus and full protective clothing. Drums should be wet down with water and removed from area of fire at the first opportunity. Equipment should be thoroughly decontaminated after use.

**HEALTH:** Results of laboratory studies with animals indicate that HEX is an extremely irritating and corrosive substance to the eye. Contact with eyes may result in eye injury or permanent damage. Studies also revealed that HEX is highly toxic through inhalation and can be absorbed through the skin to produce toxic effects.
HEX is considered as a primary skin irritant. The product may produce an allergic reaction in some humans, according to the results of a dermal sensitizing study with guinea pigs.

**PRECAUTIONS FOR NORMAL USE**

In operations employing basic, sound industrial hygiene practices, acute effects from over-exposure to HEX are not likely to occur.

To avoid adverse health effects from chronic over-exposure, the following specific practices should be employed:

A. The toxicity information, signs of over-exposure, handling precautions and use of personal protective equipment should be reviewed with each employee engaged in the manufacture or handling of HEX.

B. Employees should be required to shower if a spill or leak has caused gross contamination of work clothes or body surfaces. All employees are encouraged to shower at the end of each work shift.

C. Eating, drinking and smoking should be prohibited in the production area and employees should be instructed to thoroughly wash all exposed skin surfaces prior to these activities. No food, drink or tobacco products should be stored or carried in the production area.

**PERSONAL PROTECTIVE EQUIPMENT**

The recommended personal protective equipment shown below should be worn when handling this material unless deemed unnecessary by an industrial hygienist. Inhalation and skin contact are expected to be the primary routes of occupational exposure to HEX.

<table>
<thead>
<tr>
<th></th>
<th>Eyes</th>
<th>Skin</th>
<th>Respiratory</th>
</tr>
</thead>
<tbody>
<tr>
<td><strong>Closed Systems:</strong></td>
<td>Chemical splash goggles</td>
<td>Coveralls, impervious gloves, boots</td>
<td>None</td>
</tr>
<tr>
<td><strong>Routine Handling:</strong></td>
<td>Eye protection is provided by facepiece respirator</td>
<td>Coveralls, impervious gloves, boots</td>
<td>Full facepiece approved organic vapor respirator</td>
</tr>
<tr>
<td><strong>Gross Contact Possible:</strong></td>
<td>Eye protection is provided by facepiece respirator</td>
<td>Coveralls, impervious gloves, boots</td>
<td>Full facepiece air-supplied respirator</td>
</tr>
</tbody>
</table>

**TERMS**

*Closed Systems:* Bulk handling and storage, continuous reactors, etc. in well-ventilated areas.

*Routine Handling:* Operations requiring open weighing, transferring and routine manufacturing and packaging.

*Gross Contact Possible:* Non-routine hand operations, emergency repairs, spills and temporary ventilation failure.
EXPOSURE LIMIT

The Threshold Limit Value (TLV) of 0.01 ppm (approximately 0.1 mg/M³) was recommended by the American Conference of Governmental Industrial Hygienists (ACGIH).

Signs of over-exposure to HEX may cause one or more of the following symptoms:

- Severe eye irritation
- Marked skin irritation
- Tearing
- Salivation
- Difficulty breathing

RESULTS OF TOXICITY STUDIES

For a listing of adverse effects observed during studies, please refer to the Supplemental Toxicity Information Bulletin.

Ingestion: Acute oral toxicity studies with rats indicated that HEX is “moderately toxic” through ingestion (LD₅₀ = 584 mg/kg). [Based on Hodge-Sterner’s Acute Toxicity Scale.]

Eye: Results of a primary eye irritation study with rabbits indicated that HEX is extremely irritating and corrosive to the eye. It may also produce systemic toxic effects via this exposure route. Due to the toxicity of the substance, all test animals died during the study.

Dermal: Results of an acute dermal toxicity study with rabbits indicated that HEX can be absorbed through the skin to produce toxic effects (LD₅₀ <200). HEX is classified as a primary skin irritant (irritation score 6.1/8.0) based on the results of a primary skin irritation study with rabbits. A dermal sensitization study with guinea pigs revealed that HEX may cause an allergic reaction in some humans.

Inhalation: HEX is toxic through inhalation, according to the results of an acute inhalation study (LD₅₀ <2 mg/M³) with rabbits. Test animals showed signs of respiratory tract irritation.

Other: Mutagenicity testing results were negative for HEX. In teratology studies, HEX treated animals displayed development similar to those in the control group.

FIRST AID

Ingestion: If swallowed, DO NOT induce vomiting. If conscious, drink several glasses of water or milk. Never give anything by mouth to an unconscious person. Get prompt medical attention.

Eye: Immediately flush eyes with plenty of water for at least 15 minutes. Get prompt medical attention.

Dermal: Flush skin with plenty of water for at least 15 minutes. Remove contaminated clothing and shoes. Get prompt medical attention.

Inhalation: POISON. Call a poison control center. If overcome, remove worker to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. Get prompt medical attention.
HEX HANDLING, TRANSFER & STORAGE

The handling and storage of Hex requires special consideration in the selection of equipment materials of construction and compliance with U.S. Department of Transportation hazardous material transportation regulations (for transport). Between the temperatures of 10°C and 50°C Hex can be satisfactorily handled in 316SS and several types of plastic resin coated carbon steel.

**Piping and Valves:** 316SS pipe with welded and flanged joints is the material of choice. 316SS ball or plug valves with Teflon seat and seals are recommended. The preferred gasket material for Hex service is Garlock 7705

**Pumps & Seals:** 316SS centrifugal pumps or 316SS mag-drive seal-less pumps are recommended for Hex service.

**Storage – Tank:** Hex is stored in lined steel storage tanks. The recommended lining for large storage tanks is Coroline 505 series multi-layered modified epoxy lining with glass cloth reinforcing.

Temperature controls are necessary on Hex storage to maintain the Hex temperature above its freezing point of 10°C.

Nitrogen padding must be used on Hex storage tanks to prevent the gradual air oxidation of the Hex and resulting loss of assay. All Hex storage should be located within secure, diked, spill containment pads.

**Hex Containers (Un-Loading) - Isotanks:** Isotainers must be unloaded either by Nitrogen padding alone, or assisted by a pump. Nitrogen must be used in either case. Air should never be used as the resulting reaction with the product will both degrade the product and damage the Isotank. Pad pressure used during the unloading process must be set well below the pressure at which both the rupture disk and/or the relief valve are set to relieve.
## REGULATORY INFORMATION

### Europe

| EINECS Status | Listed, Hexachlorocyclopentadiene
|---------------|-----------------------------------------------------------------
| EC #          | 201-029-3                                                         |
| EU Classification for Dangerous Preparation Directive | Annex I Index # 602-078-00-7                                     |

### Americas

| U.S. TSCA Inventory | Listed, 1,3-cyclopentadiene, 1,2,3,4,5,5-hexanchloro
|---------------------|-----------------------------------------------------------------
| CASRN               | 77-47-4                                                           |
| Canadian DSL        | Listed                                                             |

### Pacific Rim

<table>
<thead>
<tr>
<th>Australian Inventory AICS</th>
<th>Listed</th>
</tr>
</thead>
<tbody>
<tr>
<td>Japan MITI/MHW Chemical Substances Control Inventory for Existing and New Chemical Substances</td>
<td>Listed, ENCS # (3)-2253</td>
</tr>
<tr>
<td>South Korean Existing Chemicals List</td>
<td>Listed, Korean Gazette # KE-18409</td>
</tr>
<tr>
<td>Philippines Inventory (PICCS)</td>
<td>Listed</td>
</tr>
<tr>
<td>Chinese Inventory (Draft)</td>
<td>Listed, Chinese Section #: III</td>
</tr>
</tbody>
</table>

The information in this data sheet is, to the best of our knowledge, true and accurate. The representations about the product are based upon test results achieved under laboratory practices supervised and controlled by Velsicol Chemical LLC. Since formulations, preparation or conditions of use may vary, Velsicol is unable to guarantee the same performance as indicated.

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